ETHYLENIC ACYL CYANIDES III<sup>†</sup> : BASE INDUCED SELF-CONDENSATION OF ETHYLENIC ACYL CYANIDES.

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<u>Summary</u> : Tertiary amines promote formation of ethylenic acyl cyanide enolates which add with high chemo-selectivity on acyl cyanides giving unsaturated  $\delta$ -lactones.

In connection with a project on the synthesis of a bicyclic enedione as  $\underline{3}$ , we tried to react crotonyl cyanide enolate  $\underline{2}$  (methyl crotonate  $d^5$ -reagent (2)) with cyclohexenone. However, we were unsuccessful using a variety of experimental conditions which might reasonably be expected to have given  $\underline{3}$ . For example, with potassium t-butoxide in t-butanol solution, cyclohexenone is recovered with new products which result from enolate addition on crotonyl cyanide :



The highest yield for this self-condensation (87 %) is observed under exceptionally mild conditions : pyridine/THF;  $0^{\circ}$  r.t.; 3 days; 4/5 : 15/85 (3) :



A mechanistic scheme that accommodates these, and other observations discussed below, is so illustrated ( $d^3$ -reagent on  $a^4$ -reagent (2)): <sup>†</sup>Part II : see ref. 1.



This base induced reaction can take place in DMSO with a catalytic amount of KF (4). In this case, pyranone <u>6</u> is the sole product (90 %) (during storage, the lactones <u>4</u> and <u>5</u> are isomerised to very large white crystals of the pyrone <u>6</u>):



Even when the above reaction was carried out with 1 mol equivalent of crotonyl cyanide and 10 mol equivalents of methyl vinyl ketone ( with pyridine/THF ), the major products are  $\underline{4}$  and  $\underline{5}$ . Only small quantities of other, as yet unidentified, condensation products were formed.

The enclate  $\underline{2}$  can be trapped by condensation with acetyl chloride (2 mol equivalents, pyridine/THF) giving 2-acetoxy 2,4-pentadiennitrile  $\underline{7}$  (30-35 %; E/Z  $\approx$  1/1 (5)(7)), but  $\underline{4}$  and  $\underline{5}$  are still the major products (60 %):



Condensation with trimethylchlorosilane leads to the known 2-(trimethylsilyloxy)-2,4-pentadiennitrile  $\underline{8}(8)$  also in low yield (20-25 %)( $\underline{4}, \underline{5}$ , yield : 60 %) :



Hydrolysis (basic medium) of the lactones  $\underline{4}$  and  $\underline{5}$  lead quantitatively to glutaric acid derivative  $\underline{9}$  :



Comparable results are also obtained with the senecioyl cyanide <u>10</u>. Condensation in presence of pyridine, triethylamine or KF/DMSO leads to a mixture of lactones <u>11</u> and <u>12</u> (overall yield : 80 % ; with pyridine or triethylamine, <u>11/12</u> : 40/60 ; with KF/DMSO, <u>11/12</u> : 65/35) :



Attempts to obtain 2-acetoxy-4-methyl-2,4-pentadien-

nitrile <u>13</u> by condensation with acetyl chloride (2 mol equivalent, pyridine or triethylamine/THF) lead to a complex mixture of <u>13</u> (yield 20-25 % -with pyridine-; 15 % -with triethylamine-; one isomer), <u>11+12</u> (yield 60 % -with pyridine-; 10 % -with triethylamine-), and a new self-condensation product <u>14</u> (yield 10 % -with pyridine-; 65 % -with triethylamine-) (overall yield : 85-90 %) :



The formation of the lactone <u>14</u> can be explained by a 1,2-addition of enolate on senecicyl cyanide ( $d^5$ -reagent on  $a^2$ -reagent (2)(9)) (a slight difference in the solvent used can lead to a modification of 1,2 versus 1,4 addition, for an example see (12)) :



These observations show a high chemo-selectivity and confirm that ethylenic acyl cyanides are very good acceptors (1)(13)(14).

In addition, we have observed during the preparation

of methacryloyl cyanide <u>15</u> a rather special DIELS-ALDER reaction. During the condensation of methacryloyl chloride with cuprous cyanide (15), the major product ( yield : 60 % ) is the pyran <u>16</u> resulting from the loss of CO + HCN :



Hydrolysis ( 20 % hydrochloric acid ) of the pyran <u>16</u> leads quantitatively to the  $\gamma$ -keto-acid 17 :



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- 3 Anhydrous pyridine (25 ml) and anhydrous THF (10 ml) were cooled ( 0°C ) under nitrogen and acyl cyanide ( 10 mmol ) in anhydrous THF (15 ml) was slowly added. The solution was allowed to stir at r.t. for 3 days. Hydrolysis and extractive work-up ( acidic ) with diethyl ether gave lactones.
- 4 Potassium fluoride (60 mg), DMSO (25 ml) and THF (5 ml) were cooled ( 0°C ) under nitrogen and acyl cyanide ( 10 mmol ) in anhydrous THF (10 ml) was slowly added. The solution was allowed to stir at r.t. for 3 h. Hydrolysis and extractive work-up with diethyl ether gave lactones.
- 5 l-Acetoxy-l-cyano-l,3-butadiene  $\underline{7}$  had the potential of serving as vinylketene equivalent in DIELS-ALDER reaction (vinylketenes add on the ethylenic compounds according to  $\pi 2s + \pi 2a$  process (6).
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